

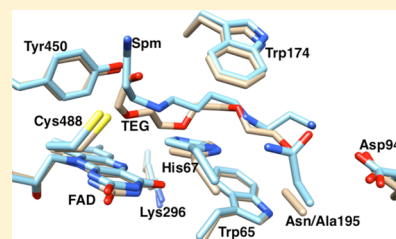
Mechanistic and Structural Analyses of the Roles of Active Site Residues in Yeast Polyamine Oxidase Fms1: Characterization of the N195A and D94N Enzymes

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ABSTRACT: Flavoprotein Fms1 from *Saccharomyces cerevisiae* catalyzes the oxidation of spermine in the biosynthetic pathway for pantothenic acid. The same reaction is catalyzed by the mammalian polyamine and spermine oxidases. The active site of Fms1 contains three amino acid residues positioned to interact with the polyamine substrate, His67, Asn195, and Asp94. These three residues form a hydrogen-bonding triad with Asn195 being the central residue. Previous studies of the effects of mutating His67 are consistent with that residue being important both for interacting with the substrate and for maintaining the hydrogen bonds in the triad [Adachi, M. S., Taylor, A. B., Hart, P. J., and Fitzpatrick, P. F. (2012) *Biochemistry* 51, 4888–4897]. The N195A and D94N enzymes have now been characterized to evaluate their roles in catalysis. Both mutations primarily affect the reductive half-reaction. With *N*¹-acetylspermine as the substrate, the rate constant for flavin reduction decreases ~450-fold for both mutations; the effects with spermine as the substrate are smaller, 20–40-fold. The $k_{\text{cat}}/K_{\text{amine}}$ – and k_{cat} –pH profiles with *N*¹-acetylspermine are only slightly changed from the profiles for the wild-type enzyme, consistent with the pK_{a} values arising from the amine substrate or product and not from active site residues. The structure of the N195A enzyme was determined at a resolution of 2.0 Å. The structure shows a molecule of tetraethylene glycol in the active site and establishes that the mutation has no effect on the protein structure. Overall, the results are consistent with the role of Asn195 and Asp94 being to properly position the polyamine substrate for oxidation.



The polyamines spermidine and spermine are polybasic molecules that are ubiquitous in living organisms.¹ Because polyamines are required by growing cells, the enzymes of polyamine metabolism have been targets for the development of anticancer drugs.^{2,3} Catabolism of polyamines in mammals occurs by one of two pathways.⁴ Both spermine and spermidine can be acetylated on a terminal nitrogen by spermidine/spermine-*N*¹-acetyltransferase. The *N*¹-acetylspermine or *N*¹-acetylspermidine can then be oxidized by the peroxisomal flavoprotein polyamine oxidase (PAO, also known as *N*-acetylpolyamine oxidase) to produce spermidine or putrescine, respectively, and *N*-acetyl-3-aminopropanaldehyde (Scheme 1). Alternatively, spermine and spermidine can be oxidized directly by the separate flavoprotein spermine oxidase (SMO) to spermidine or putrescine and 3-aminopropanaldehyde. Plants and yeast also contain flavoproteins capable of catalyzing polyamine oxidation. *Saccharomyces cerevisiae* protein Fms1 oxidizes spermine and *N*¹-acetylspermine like the mammalian enzymes.^{5,6} The 3-aminopropanal produced by Fms1 is further metabolized to β -alanine, a precursor of pantothenic acid.^{5,6} The plant enzymes oxidize spermine and spermidine and their acetylated derivatives on the *endo* side of a nitrogen, producing 1,3-diaminopropane and the aldehyde form of *N*¹-acetyl-spermidine from *N*¹-acetylspermine.⁷

The structural bases for the different substrate specificities of the polyamine-oxidizing flavoproteins are not known. Structures of maize PAO⁸ and Fms1⁹ are available, but not that of any mammalian SMO or PAO. These structures establish that the maize enzyme and Fms1 belong to the monoamine oxidase structural family.¹⁰ The sequences of the mammalian SMOs and PAOs, while only ~20% identical to those of maize PAO and Fms1, indicate that they also have the same fold as monoamine oxidase. Because the maize enzyme catalyzes a reaction different from that of Fms1 and the mammalian enzymes, Fms1 is at present the best model for understanding the structural basis for catalysis and specificity of the mammalian polyamine-oxidizing enzymes.

The structure of Fms1 shows that the active site contains three polar residues capable of interacting with a polyamine substrate, His67, Asn195, and Asp94. On the basis of a structure of Fms1 in the presence of spermine, these three residues have been proposed to form hydrogen bonds with the nitrogens of the amine and with one another (Figure 1).^{9,a} His67, which is conserved in the mammalian enzymes, would form a hydrogen bond with N4 of the substrate, the site of oxidation, and with Asn195. Asn195 would form a hydrogen bond with N12 of spermine and with Asp94, which in turn also interacts with N12 of the substrate. Mechanistic and structural

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Scheme 1

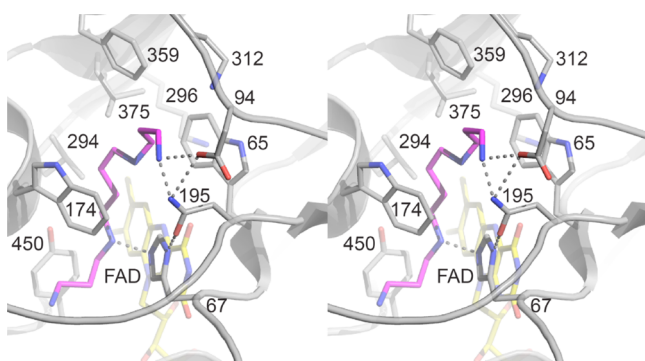
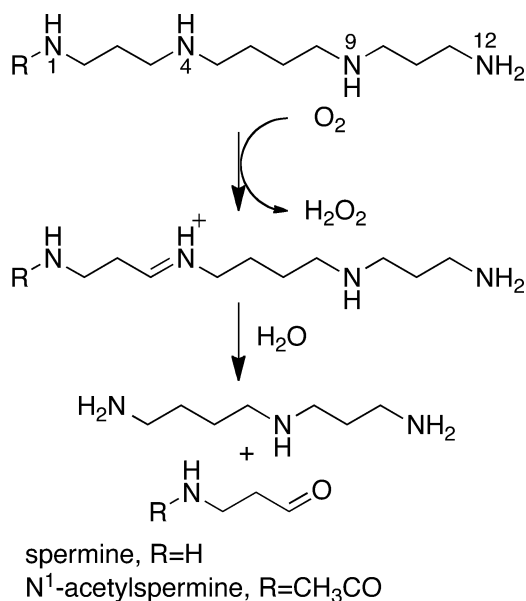


Figure 1. Active site residues in *S. cerevisiae* Fms1 showing the proposed interactions with spermine. The structure is based on subunit B of PDB entry 1XPQ.

studies of the effects of mutating His67 of Fms1¹¹ and His64 of mouse PAO¹² suggest that this histidine residue plays a role both in properly positioning the substrate for hydride transfer to the flavin and in maintaining the position of Asn195. In this study, we describe the effects of replacing Asn195 of Fms1 with alanine and Asp94 with asparagine.

MATERIALS AND METHODS

Materials. The pET28-based pJWL94 vector encoding histidine-tagged Fms1⁶ was kindly provided by R. Sternglanz (Stony Brook University, Stony Brook, NY). *Escherichia coli* BL21(DE3) Codon⁺ RIL competent cells and the QuikChange site-directed mutagenesis kit were from Stratagene (Santa Clara, CA). The QIAprep Spin Miniprep kit was from Qiagen (Valencia, CA). Oligonucleotides used for site-directed mutagenesis and for sequencing of the mutant genes were synthesized by the Nucleic Acid Core Facility at the University of Texas Health Science Center at San Antonio. Kanamycin, chloramphenicol, isopropyl β -D-thiogalactopyranoside, lysozyme, Luria-Bertani agar and broth, phenylmethanesulfonyl fluoride, and HEPES were from Fisher (Pittsburgh, PA). N¹-Acetylspermine trihydrochloride and glucose oxidase were from Sigma-Aldrich (Milwaukee, WI). Spermine was purchased from

Acros Organics (Geel, Belgium). The nickel-nitrilotriacetic acid agarose and the Sephacryl S-100 HR resin were purchased from Invitrogen (Carlsbad, CA) and Sigma-Aldrich, respectively.

Site-Directed Mutagenesis. The mutations of residues Asn195 and Asp94 were introduced into the gene encoding wild-type Fms1⁶ using the QuikChange protocol (Stratagene). The oligonucleotides used as forward primers, with the mutations underlined, were 5'-CACCAAG-GAAGGCGCCCTTTGCTTTG-3' and 5'-GTTTTGA-TAACGATAATTTTATTATATCGACG-3' for the N195A and D94N mutations, respectively. The entire sequences of the mutant genes were determined at the Nucleic Acid Core Facility at the University of Texas Health Science Center at San Antonio.

Protein Expression and Purification. Wild-type and mutant Fms1 were expressed and purified following the protocol for the wild-type enzyme.¹³ The levels of protein expression and the yields of the purified mutant proteins were similar to those for wild-type Fms1. For crystallization, the N195A enzyme was purified further using a Sephacryl S-100 column as described previously.¹¹ The purified protein was dialyzed against three changes of 25 mM HEPES, 25 mM NaCl, and 2% glycerol (pH 7.5) before crystallization.

Enzyme Assays. The enzymatic activities of both N195A and D94N Fms1 were determined in buffer containing 10% glycerol by following oxygen consumption with a Yellow Springs Instrument model 5300 oxygen monitor at 25 °C. The steady-state kinetic parameters k_{cat} , K_{amine} , and $k_{\text{cat}}/K_{\text{amine}}$ were determined at an oxygen concentration of 1.2 mM and varying concentrations of spermine (0.05–3 mM) at pH 9.35 or N¹-acetylspermine (0.05–2 mM) at pH 9.0. The parameters K_{O_2} and $k_{\text{cat}}/K_{\text{O}_2}$ were determined by varying the concentration of oxygen (0.065–1.2 mM) at 20 mM spermine at pH 9.35 or 20 mM N¹-acetylspermine at pH 9.0. The reaction mixture (1 mL) was first equilibrated with the substrates at the desired concentrations, and the appropriate O₂/N₂ mixture was bubbled into the cell of the oxygen electrode for 10 min prior to starting the reaction by adding enzyme. The buffers were 200 mM bis-Tris at pH 6.8, 200 mM Tris-HCl from pH 7.0 to 8.75, 200 mM CHES from pH 9.0 to 9.75, and 200 mM CAPS from pH 10.1 to 10.5. Because of the hydroscopic nature of N¹-acetylspermine, its concentration was determined enzymatically.

Rapid-reaction analyses were conducted using an Applied Photophysics SX-20MV stopped-flow spectrophotometer in the absorbance mode. The instrument was thermostated at 25 °C, and enzyme and substrate solutions were made anaerobic as described previously.¹³ Enzyme (~20 μ M) was mixed anaerobically with varying concentrations of spermine (0.03–2 mM) or N¹-acetylspermine (0.03–1 mM) in 200 mM CHES (pH 9.0), and the reaction was followed at 458 nm.

Data Analysis. Kinetic data were analyzed using KaleidaGraph (Adelbeck Software, Reading, PA). The steady-state kinetic parameters k_{cat} , $k_{\text{cat}}/K_{\text{M}}$, and K_{M} were determined by fitting the data to the Michaelis–Menten equation. The $k_{\text{cat}}/K_{\text{amine}}$ values at different pH values for both N195A and D94N Fms1 were fit to eq 1

$$\log Y = \log \left(\frac{C}{1 + \frac{H}{K_1} + \frac{K_2}{H}} \right) \quad (1)$$

where K_1 and K_2 are the dissociation constants for the ionizable groups and C is the pH-independent $k_{\text{cat}}/K_{\text{amine}}$ value. The k_{cat} values for the mutant enzymes as a function of pH were fit to eq 2

$$\log Y = \log \left(\frac{Y_L + Y_H \frac{K_1}{H}}{1 + \frac{K_1}{H}} \right) \quad (2)$$

where K_1 is the dissociation constant for the ionizable group, Y_L is the activity at low pH, and Y_H is the activity at high pH.¹⁴ Stopped-flow traces were fit to eq 3, which describes a biphasic exponential process

$$A = A_{\infty} + A_1 e^{-\lambda_1 t} + A_2 e^{-\lambda_2 t} \quad (3)$$

where λ_1 and λ_2 are the first-order rate constants for each phase, A_1 and A_2 are the absorbances of each species at time t , and A_{∞} is the absorbance at infinite time. Kinetic parameters for the reductive half-reaction were determined using eq 4

$$k_{\text{obs}} = \frac{k_2 S}{K_d + S} \quad (4)$$

where k_{obs} is the observed first-order rate constant for the reduction of the enzyme-bound flavin at a specific concentration of the amine, k_2 is the rate constant for reduction at saturating concentrations of the amine, and K_d is the apparent dissociation constant for binding of the amine to the enzyme. An alternative fit, in which a non-zero y -intercept was included, did not improve the quality of the fits to the data and yielded values for the y -intercepts not significantly different from zero.

Crystallization, Structure Determination, and Refinement. Crystals of Fms1 N195A were grown in the UTHSCSA X-ray Crystallography Core Laboratory using the Morpheus crystallization screen kit (Molecular Dimensions Ltd., Newmarket, U.K.) and a Phoenix crystallization robot (Art Robbins Instruments, Sunnyvale, CA). The crystals grew within 1 week by the sitting drop vapor diffusion method with the protein solution mixed in a 1:1 ratio with buffer containing 30 mM diethylene glycol, 30 mM triethylene glycol, 30 mM tetraethylene glycol, 30 mM pentaethylene glycol, 10% ethylene glycol, 20% polyethylene glycol 8000, and 0.1 M 2-(*N*-morpholino)ethanesulfonic acid (MES)-imidazole (pH 6.5). Data were collected from a crystal flash-cooled with liquid nitrogen on beamline 24-ID-C at the Advanced Photon Source (Argonne National Laboratory, Argonne, IL). The data were integrated and scaled using HKL-2000.¹⁵ Initial phases were obtained by the molecular replacement method implemented in PHASER¹⁶ using Fms1 coordinates in PDB entry 4ECH¹¹ as the search model. Coordinates were refined against the data using PHENIX,¹⁷ including simulated annealing, alternating with manual rebuilding using COOT.¹⁸ Data collection and refinement statistics are listed in Table 1. The coordinates have been deposited in the PDB as entry 4GDP.

RESULTS

Steady-State Kinetic Parameters. To analyze the role of Asn195 and Asp94 of Fms1 in binding and catalysis, Asn195 was mutated to alanine and Asp94 was mutated to asparagine. The steady-state kinetic parameters of the mutant enzymes with spermine and *N*¹-acetylspermine as substrates are given in Table 2. The analyses were conducted at the respective pH optima for the wild-type enzyme, 9.0 for *N*¹-acetylspermine and 9.35 for spermine.¹³ The N195A mutation results in a decrease

Table 1. Data Collection and Refinement Statistics

Data Collection	
space group	P1
cell dimensions	
<i>a</i> , <i>b</i> , <i>c</i> (Å)	78.9, 81.9, 104.8
α , β , γ (deg)	78.1, 79.5, 78.9
wavelength (Å)	0.97920
resolution (Å)	100.0–2.0
<i>R</i> _{sym}	0.109 (0.408) ^a
<i>I</i> / σ <i>I</i>	10.5 (3.2) ^a
completeness (%)	89.0 (75.3) ^a
redundancy	3.4 (3.2) ^a
Refinement	
resolution (Å)	79.1–2.0
no. of reflections	149579
<i>R</i> _{work} / <i>R</i> _{free}	0.203/0.234
no. of monomers per unit cell	4
no. of atoms	
protein	15909
ligand	260
solvent	723
<i>B</i> factor (Å ²)	
protein	34.0
ligand	25.4
solvent	36.3
rmsd	
bond lengths (Å)	0.004
bond angles (deg)	0.876

^aValues in parentheses are for the highest-resolution shell.

in the $k_{\text{cat}}/K_{\text{amine}}$ value of ~ 1 order of magnitude with both amine substrates, with a slightly larger effect on *N*¹-acetylspermine oxidation. The effect of this mutation on the k_{cat} value with spermine mirrors the effect on the $k_{\text{cat}}/K_{\text{M}}$ value, while the k_{cat} value with *N*¹-acetylspermine decreases only ~ 2 -fold. Mutating Asn195 to alanine has no significant effect on the $k_{\text{cat}}/K_{\text{M}}$ value for oxygen with either substrate. These results are consistent with this mutation primarily affecting the reductive half-reaction.

The D94N mutation has a much larger effect on the $k_{\text{cat}}/K_{\text{amine}}$ and k_{cat} values for spermine than those for *N*¹-acetylspermine (80- vs 4-fold and 20- vs 1.5-fold, respectively). With either *N*¹-acetylspermine or spermine as the polyamine substrate, the $k_{\text{cat}}/K_{\text{O}_2}$ value decreases ~ 4 -fold for the D94N enzyme. Thus, this mutation affects both the reductive and oxidative half-reactions, with a larger effect on the former.

pH Dependence of the k_{cat} and $k_{\text{cat}}/K_{\text{amine}}$ Values. To gain further insight into the basis for the changes in the $k_{\text{cat}}/K_{\text{M}}$ and k_{cat} values, $k_{\text{cat}}/K_{\text{amine}}$ – and k_{cat} –pH profiles were determined with *N*¹-acetylspermine as the amine substrate for the mutant enzymes. For both enzymes, the $k_{\text{cat}}/K_{\text{amine}}$ –pH profiles are bell-shaped, with maxima of ~ 9.0 , similar to the profile for the wild-type enzyme (Figure 2A). These pH profiles show the importance of two ionizable groups in the free enzyme or substrate, one of which must be protonated and one unprotonated for full activity. The data were fit to eq 1 to extract the $\text{p}K_{\text{a}}$ values (Table 3). For the N195A enzyme, the two $\text{p}K_{\text{a}}$ values are sufficiently separated to yield discrete $\text{p}K_{\text{a}}$ values that are unchanged from the values for the wild-type enzyme. For the D94N enzyme, the two $\text{p}K_{\text{a}}$ values are too close together to resolve, so that only the average $\text{p}K_{\text{a}}$ value of the two ionizable groups can be determined; this value is not

Table 2. Steady-State Kinetic Parameters for Wild-Type Fms1 and the N195A and D94N Enzymes^a

substrate	kinetic parameter	Fms1 ^d	N195A	D94N
<i>N</i> ¹ -acetylspermine ^b	k_{cat} (s ⁻¹) ^c	15.1 ± 0.4	8.1 ± 0.2	9.7 ± 0.2
	$k_{\text{cat}}/K_{\text{amine}}$ (mM ⁻¹ s ⁻¹) ^c	1400 ± 200	78 ± 13	320 ± 50
	K_{amine} (μM) ^c	10.9 ± 1.8	104 ± 14	30 ± 5
	$k_{\text{cat}}/K_{\text{O}_2}$ (mM ⁻¹ s ⁻¹) ^e	358 ± 20	405 ± 72	100 ± 14
	K_{O_2} (μM) ^e	43.6 ± 2.3	20 ± 3	97 ± 14
spermine ^f	k_{cat} (s ⁻¹) ^c	39.0 ± 1.5	4.9 ± 0.2	2.0 ± 0.3
	$k_{\text{cat}}/K_{\text{amine}}$ (mM ⁻¹ s ⁻¹) ^c	330 ± 60	38.5 ± 8.3	4.1 ± 1.0
	K_{amine} (μM) ^c	118 ± 25	127 ± 27	320 ± 88
	$k_{\text{cat}}/K_{\text{O}_2}$ (mM ⁻¹ s ⁻¹) ^g	428 ± 77	327 ± 67	104 ± 36
	K_{O_2} (μM) ^g	91 ± 16	15 ± 3	192 ± 60

^aAt 25 °C. ^bDetermined at pH 9.0. ^cDetermined by varying the concentration of the amine at 1.2 mM oxygen. ^dFrom ref 13. ^eDetermined by varying the concentration of oxygen at 20 mM *N*¹-acetylspermine. ^fDetermined at pH 9.35. ^gDetermined by varying the concentration of oxygen at 20 mM spermine.

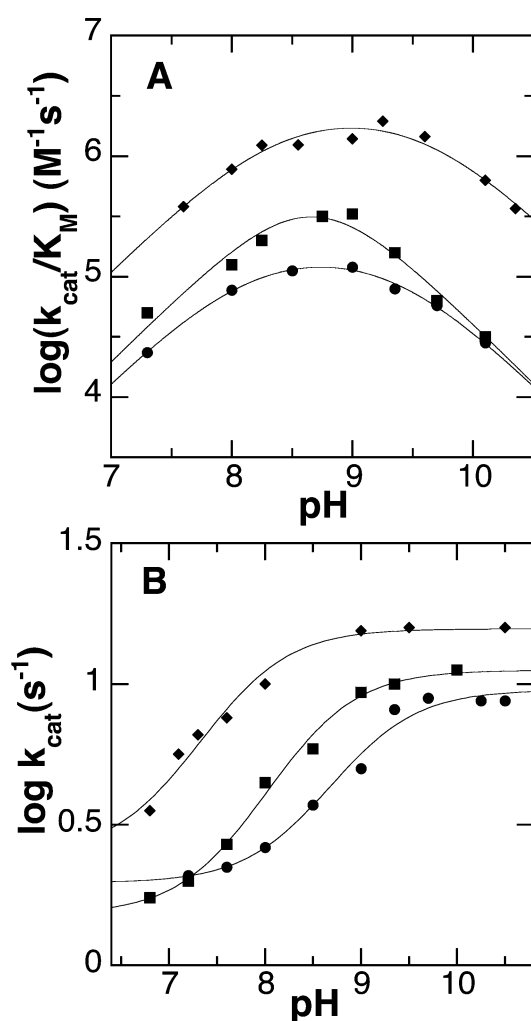


Figure 2. pH profiles for Fms1 mutant enzymes. (A) Effect of pH on the $k_{\text{cat}}/K_{\text{amine}}$ value with *N*¹-acetylspermine as the substrate for wild-type (◆), D94N (■), and N195A (●) Fms1. The lines are from fits of the data to eq 1. (B) Effect of pH on the k_{cat} value with *N*¹-acetylspermine for wild-type (◆), D94N (■), and N195A (●) Fms1. The lines are from fits to eq 2. For the mutant enzymes, the concentration of *N*¹-acetylspermine was varied at an oxygen concentration of 1.2 mM at 25 °C. The data for the wild-type enzyme are from ref 13.

Table 3. pK_a Values for Wild-Type Fms1 and the D94N and N195A Enzymes^a

enzyme	eq	kinetic parameter	pK ₁	pK ₂
wild-type Fms1 ^b	1	$k_{\text{cat}}/K_{\text{amine}}$	8.3 ± 0.1	9.6 ± 0.1
	2	k_{cat}	7.7 ± 0.1	—
N195A	1	$k_{\text{cat}}/K_{\text{amine}}$	8.1 ± 0.1	9.4 ± 0.2
	2	k_{cat}	9.0 ± 0.1	—
D94N	1	$k_{\text{cat}}/K_{\text{amine}}$	8.7 ± 0.1	8.7 ± 0.1
	2	k_{cat}	8.4 ± 0.1	—

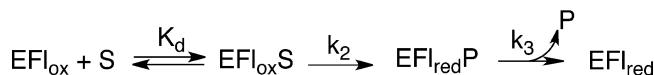
^aConditions: *N*¹-acetylspermine as a substrate, 25 °C. ^bFrom ref 13.

significantly different from the average of the pK_a values for the wild-type enzyme (9.0 ± 0.2).

The effects of pH on the k_{cat} values for both mutant enzymes with *N*¹-acetylspermine as the amine substrate are shown in Figure 2B. As is the case for the wild-type enzyme, the k_{cat} values for the mutant enzymes increase with pH to limiting values at high pH, consistent with the requirement for an unprotonated group for the fastest turnover of both the wild-type and mutant enzymes. To obtain the pK_a values (Table 3), the data were fit to eq 2. For both mutant enzymes, this pK_a value is higher than that seen for the wild-type enzyme. As a result, while the k_{cat} value for the wild-type enzyme is constant at pH ≥ 9.0, the k_{cat} values for the mutant enzymes do not reach a maximum until pH ≥ 10. The data in Figure 2B can be used to obtain the pH-independent k_{cat} values for the mutant enzymes using eq 2. This yields k_{cat} values of 11.5 ± 0.6 s⁻¹ for the N195A enzyme and 12.7 ± 0.6 s⁻¹ for the D195N enzyme; both are very close to the wild-type value of 15 s⁻¹.

Rapid-Reaction Kinetics. Rapid-reaction methods were used to determine directly the effects of the mutations on the rate constant for flavin reduction. The N195A and D94N enzymes were mixed with *N*¹-acetylspermine or spermine anaerobically in a stopped-flow spectrophotometer at pH 9.0 and 25 °C, and the changes in the visible absorbance spectrum of the FAD were followed. As was the case with the wild-type enzyme,¹³ the decrease in absorbance was biphasic with most of the absorbance change occurring in the more rapid first phase. Only the rate constant for the fast phase exhibited a dependence on the concentration of the amine substrate, consistent with the mechanism of Scheme 2. The limiting rate constant for flavin reduction (k_2 in Scheme 2) and the apparent dissociation constant (K_d) for binding of the amine to the enzyme were determined by fitting the rate constant for the

Scheme 2



first phase as a function of the amine concentration to eq 4. For both mutant enzymes, the rate constant for the slow phase was independent of the amine concentration and much slower than turnover, so that it is not along the catalytic pathway. Both mutations result in significant decreases in the k_2 values with respect to the wild-type enzyme (Table 4). With N^1 -acetyl spermine as the amine substrate, the value of k_2 decreases ~600- and ~350-fold for the N195A and D94N enzymes, respectively, compared to the value for the wild-type enzyme. The apparent K_d values for the mutant enzymes are both ~8-fold lower than that for wild-type Fms1. The effect is weaker with spermine as a substrate. The value of k_2 decreases ~7- and ~20-fold while the apparent K_d value increases 4- and 12-fold for the N195A and D94N enzymes, respectively.

Crystal Structures of N195A Fms1. Crystallization screens were performed with both mutant proteins. For D94N Fms1, the best crystal diffracted to 3.7 Å, which was sufficient to establish that the mutation did not cause major changes in the overall structure of the protein (results not shown). In contrast, the structure of N195A Fms1 could be determined at a resolution of 2.0 Å (Figure 3). The protein crystallized as a tetramer, similar to the previous structures of the wild-type enzyme in the presence of spermine⁹ and of the H67Q mutant enzyme.¹¹ The structure of N195A Fms1 lacks the first six residues and residues 347 and 348, a mobile loop that is also not in seen the wild-type enzyme. The overall structure is similar to that of the wild-type enzyme, with an rmsd of 0.437 Å for 482 α -carbons versus PDB entry 1YY5. Most of the differences are in surface loops, especially residues 131–136 and 419–427; the latter is disordered in two of the subunits of the mutant protein. The loop containing residues 456–459 missing in the structure of the wild-type enzyme is observed in the mutant protein structure.

The structure of the active site is not disrupted by replacing Asn195 with alanine, in that all of the active site residues, including His67 and Asp94, occupy equivalent positions in the mutant enzyme (Figure 4). The short alanine side chain at position 195 of the mutant occupies the same position as the methylene of the native aspartate. The isoalloxazine ring of the FAD was best modeled with a slight bend of ~7°. This bend was not reported for the wild-type structures, nor was it detected in our previous description of the structure of H67Q Fms1,¹¹ all of which were determined at lower resolution than the structure presented here (2.3–2.4 Å vs 2.0 Å).⁹

Additional electron density is observed in the active site of N195A Fms1 in the same position that spermine has been reported to bind in the wild-type enzyme.⁹ This density fits well with a molecule of tetraethylene glycol (Figure 3), which was

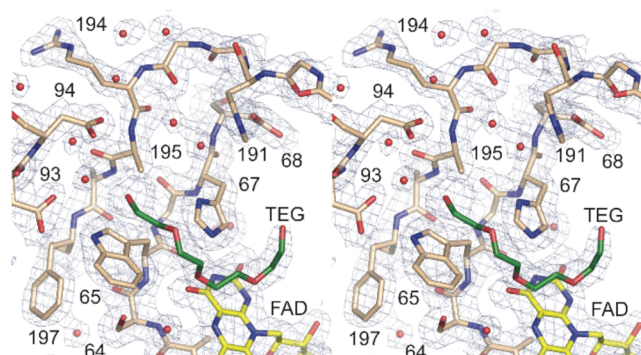


Figure 3. Refined N195A Fms1 structure (tan) superimposed onto a SIGMA-A weighted electron density map with coefficients $2mF_o - F_c$ and contoured at 1σ .

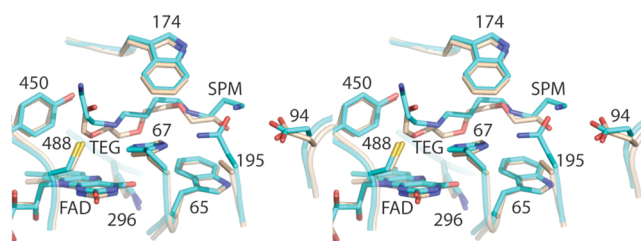


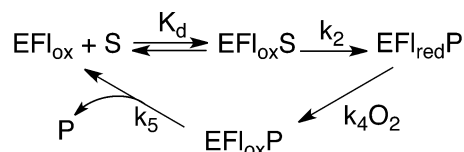
Figure 4. Comparison of active sites of wild-type (cyan) and N195A (tan) Fms1. TEG stands for tetraethylene glycol and Spm for spermine. The wild-type structure is from subunit B of PDB entry 1XPQ.

present in the crystallization solution. Tetraethylene glycol is a reasonable analogue for the substrate spermine, although it is one atom shorter. The nearest carbon of the tetraethylene glycol is 3.2 Å from imidazole N3 atom of His67, consistent with the proposed role of this residue in binding a secondary nitrogen of the substrate. A terminal oxygen of the ligand is 3.2 Å from Cys488, suggesting that a role of this residue is to bind N1 of the polyamine substrate. The other terminal oxygen is too far from Asp94 for a hydrogen bond (4.1 Å), but the additional atom of spermine could place the substrate N14 appropriately for an ionic interaction with this residue.

DISCUSSION

The kinetic mechanism for Fms1 is shown in Scheme 3.¹³ The effects of the N195A and D94N mutations on the values of the

Scheme 3


 Table 4. Rapid-Reaction Kinetic Parameters for Wild-Type Fms1 and Mutant Enzymes^a

substrate	kinetic parameter	Fms1 ^b	N195A	D94N	H67Q ^b
N^1 -acetyl spermine	K_d (μM)	484 ± 83	60 ± 10	73 ± 17	23 ± 1
	k_2 (s^{-1})	5490 ± 90	9.0 ± 0.2	15.6 ± 0.7	58 ± 3
spermine	K_d (μM)	23 ± 8	84 ± 20	282 ± 38	3300 ± 500
	k_2 (s^{-1})	126 ± 3	16.4 ± 0.6	6.10 ± 0.25	2.1 ± 0.5

^aDetermined at pH 9.0 and 25 °C. ^bFrom ref 11.

individual rate constants in the mechanism can be determined from the steady-state and rapid-reaction analyses described here. The apparent K_d value for polyamine binding and the rate constant for flavin reduction, k_2 , were measured in stopped-flow analyses of the reductive half-reaction in the absence of oxygen. The second-order rate constant for the reaction of the reduced flavin–product complex with oxygen is equal to the $k_{\text{cat}}/K_{\text{O}_2}$ value determined in steady-state kinetic analyses.¹³ With both mutant enzymes, the effects of the mutation are greater on the reductive half-reaction than on the oxidative half-reaction. This resembles the effects of mutating His67.¹¹ Mechanistic studies of amine-oxidizing flavoproteins, including the polyamine-oxidizing enzymes, are most consistent with reaction involving direct transfer of a hydride from the uncharged carbon–hydrogen bond of the substrate to the flavin cofactor,^{19–21} although other mechanisms have been proposed.²² In such a mechanism, there is no need for an active site base because catalysis requires that the substrate bind with the reacting nitrogen in the unprotonated form.^{23–25} Instead, the role of active site residues is to properly position the substrate for oxidation. These results are consistent with such a role for both Asn195 and Asp94.

Both N^1 -acetylspermine and spermine were used as substrates in these analyses. The relative $k_{\text{cat}}/K_{\text{amine}}$ values for the two substrates suggest that wild-type Fms1 has only a slight preference for the acetylated polyamine (Table 2),¹³ placing it intermediate in specificity between the mammalian PAOs and SMOs.^{23,26,27} However, the rate constant for oxidation of N^1 -acetylspermine by Fms1 is ~200-fold greater than the k_2 value for oxidation of spermine (Table 4). This difference in intrinsic rate constants is not reflected in the $k_{\text{cat}}/K_{\text{amine}}$ values because N^1 -acetylspermine has a strong forward commitment to catalysis with wild-type Fms1 because of the rate constant for dissociation of this substrate from the oxidized enzyme being smaller than the rate constant for flavin reduction.¹¹ As a result, the $k_{\text{cat}}/K_{\text{amine}}$ value for N^1 -acetylspermine with wild-type Fms1 reflects the second-order rate constant for substrate binding rather than the rate constant for the chemical step. The k_{cat} value for this substrate similarly reflects the rate constant for release of the oxidized amine from the oxidized enzyme rather than for chemistry. Thus, changes in the steady-state kinetic parameters for N^1 -acetylspermine for the mutant enzymes do not fully reflect the effects of the mutations on individual rate constants. In contrast, the more slowly oxidized substrate spermine appears to have very little commitment to catalysis, and product release is only approximately twice as fast as flavin reduction for the wild-type enzyme; therefore, changes in steady-state kinetic parameters with spermine more accurately reflect changes in intrinsic kinetic parameters.

The structure of N195A Fms1 establishes clearly that the only significant change in the mutant protein is the loss of the amide moiety of Asn195. Thus, the effects of this mutation can be attributed solely to local effects, primarily the loss of hydrogen bonds between this residue and the substrate and with both His67 and Asp94. The active site of Fms1 is a U-shaped tunnel open to the solvent at both ends and passing by the flavin isoalloxazine ring.⁹ The active site of maize PAO is similar,⁸ so that this shape is likely a common feature of PAOs and SMOs. Within the Fms1 active site, His67, Asn195, and Asp94 are appropriately placed to interact with N4 and N12 of the polyamine substrate, as shown in Figure 1. The substrate N1 atom extends up the tunnel toward the solvent. The tunnel

is lined primarily with backbone atoms. One exception is Cys488; on the basis of the position of the tetraethylene glycol in the active site of the mutant enzyme, the thiol of this residue could interact with the uncharged N1 atom of the substrate. The $k_{\text{cat}}/K_{\text{amine}}$ –pH profiles of wild-type Fms1 support a requirement that N1 and N4 be neutral for flavin reduction, while N9 and N12 must be positively charged.¹³ The residues in the Fms1 active site most likely to have $\text{p}K_a$ values between 7 and 10 are His67 and Asp94; the finding that mutating these residues does not affect the $k_{\text{cat}}/K_{\text{amine}}$ –pH profiles is consistent with this requirement.¹¹ The higher k_{cat}/K_M value for N^1 -acetylspermine is consistent with this requirement for a neutral N1 and with the lack of negatively charged amino acid residues that could interact with a positively charged N1. Interactions of the neutral acetyl group at N1 of N^1 -acetylspermine with the enzyme are thus likely to be mainly hydrophobic interactions that would not be relevant for interactions with spermine. The difference in interactions between the substrate tunnel and the substrate N1 atom and its substituents provides a reasonable explanation for the slower dissociation of the acetylated substrate and product from the enzyme.

The D94N and N195A mutations significantly alter both the K_d and k_2 values with spermine. The D94N mutation has the greater effect, increasing the K_d value for spermine by 1 order of magnitude and decreasing the rate constant for flavin reduction by 20-fold. The N195A mutation results in more modest although still significant changes in both parameters. The changes in the $k_{\text{cat}}/K_{\text{amine}}$ value for spermine reflect these changes in k_2 and K_d . The effects of both mutations on all three kinetic parameters are significantly lower than the previously reported effects of the H67Q mutation ($K_d = 3.3$ mM, and $k_2 = 2.1$ s^{−1}),¹¹ consistent with the interaction with N4, the nitrogen in the bond being oxidized, being much more critical for reactivity with this slow substrate. As noted above, the lack of an acetyl group on N1 of spermine likely precludes interactions that stabilize the binding of N^1 -acetylspermine. The magnitudes of the changes in the kinetic parameters for the D94N and N195A enzymes are consistent with the loss of the proposed interactions with the nonreacting nitrogen.

Because of the significant forward commitment to catalysis of wild-type Fms1 with N^1 -acetylspermine as the substrate, quantitative interpretation of the effects of the mutation with this substrate is more difficult. Significant forward commitments alter the K_d value obtained from rapid-reaction experiments just as they alter K_M values.^{28–30} This problem is compounded in this case by the very high k_2 value with N^1 -acetylspermine as the substrate for Fms1, which is too fast to measure at 25 °C directly and was only estimated by extrapolation from measurements at 5–15 °C.¹¹ Thus, the K_d value for the wild-type enzyme in Table 4 with this substrate is not a reliable measure of the actual equilibrium constant for binding of N^1 -acetylspermine to wild-type Fms1. Still, comparison of the k_2 and K_d values for the three mutant proteins suggests that mutagenesis of Asp94 or Asn195 has a larger effect with this faster substrate than with spermine. This may reflect more optimized interactions with N^1 -acetylspermine that are more sensitive to any slight perturbation. In addition, both mutations result in larger effects on binding and catalysis with N^1 -acetylspermine than does mutagenesis of His67 ($K_d = 23$ μM, and $k_2 = 58$ s^{−1}).¹¹ This suggests that the interactions of the N^1 -acetyl moiety with the protein compensate for the weaker interaction with N4 in the H67Q

enzyme, but any such effect is attenuated for Asn195 and Asp94, residues that interact with the more distant N12.

The D94N mutation results in a 4-fold decrease in the second-order rate constant for the reaction of oxygen with the complex of the reduced enzyme and oxidized amine, while the N195A mutation does not appear to have any effect. The rate constant for the reaction of reduced flavoproteins with oxygen is very sensitive to the presence of positive charges in the vicinity of the isoalloxazine ring; these can be supplied by the protein or the substrate.^{31–34} For all three mutations of Fms1 in Table 4, there is a correlation between the effect of the mutation on the K_d value for spermine and the effect on the k_{cat}/K_{O_2} value, with both parameters being affected the most by mutation of His67 ($k_{cat}/K_{O_2} = 28 \text{ mM}^{-1} \text{ s}^{-1}$)¹¹ and the least by mutation of N195A. If the positive charge on the oxidized amine contributes to the catalysis of the reaction of the reduced enzyme with oxygen, a disruption of the binding of the product would likely disrupt the overall electrostatics of the active site. Given the similar structures of the polyamine reactant and product (Scheme 1), mutations that alter the binding of the substrate would be expected also to alter the binding of the oxidized amine product.

The k_{cat}/K_M -pH profile for amine substrates for Fms1 has been attributed to the need for the form of the substrate in which only N8 and N12 are charged, because the pH optimum matches the pH at which the doubly charged form of the polyamine substrate is maximal.¹³ The lack of a change in the general shape or pH optimum of the k_{cat}/K_{amine} -pH profile in the mutant proteins is consistent with this pH profile reflecting the protonation state of the substrate rather than any amino acid residue in the protein. The k_{cat}/K_{amine} -pH profile does become narrower in the mutant proteins. Significant external forward commitments to catalysis will perturb pK_a values outward, resulting in a broader pH profile.³⁵ The narrower profiles with the mutant proteins can be attributed to a decrease in the commitment as the value of k_2 decreases. The k_{cat} -pH profile for wild-type Fms1 with N^1 -acetylspermine reflects the pH dependence of k_5 . The similar shapes of the k_{cat} profiles for the mutant proteins rule out Asp94 as the source of the pK_a ; His67 was similarly eliminated as a potential source by mutagenesis.¹¹ One possibility is that this pK_a reflects the protonation state of the amine product. The change in the pK_a in the two mutant proteins can be attributed to a change in the rate-determining step so that k_2 becomes closer to the rate constant for product release.

These results establish that both Asn195 and Asp94 are important for properly positioning the polyamine substrate for oxidation. The quantitative effects of mutating these residues depend on the polyamine substrate used, in that the acetyl moiety in N^1 -acetylspermine provides interactions not available to spermine. Moreover, both residues appear to be less critical than His67, a residue conserved among PAOs and SMOs.

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Notes

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ABBREVIATIONS

PAO, polyamine oxidase; SMO, spermine oxidase; PDB, Protein Data Bank.

ADDITIONAL NOTE

^aIn PDB entry 1XPQ, Asn195 is shown with the amide nitrogen interacting with His67 and the oxygen interacting with a carboxylate oxygen of Asp65. In response to a suggestion by a reviewer, we are showing the reverse orientation for Asn195, because a hydrogen bond between the amide nitrogen of Asn195 and a carboxylate oxygen of Asp94 is more chemically reasonable.

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